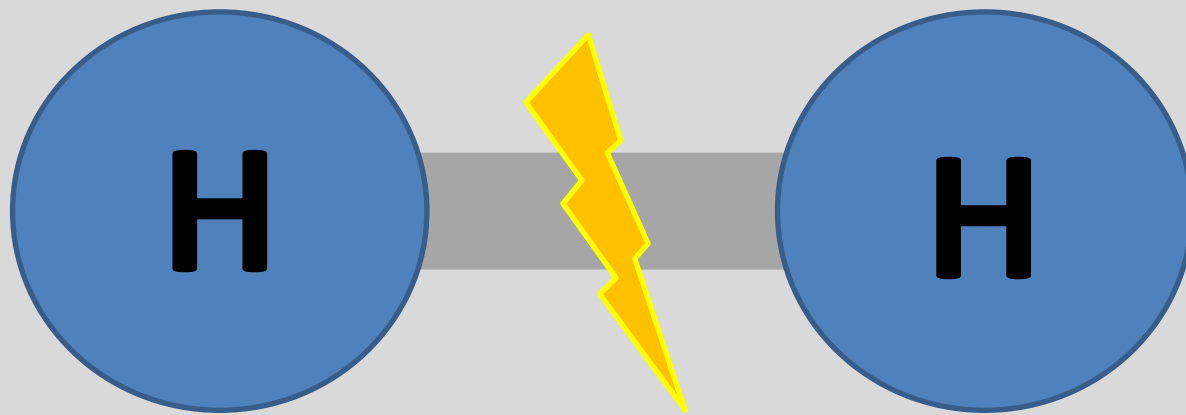


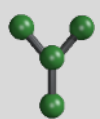
# Transition Metal Sulfide Heterostructures for Hydrogen Evolution Reaction



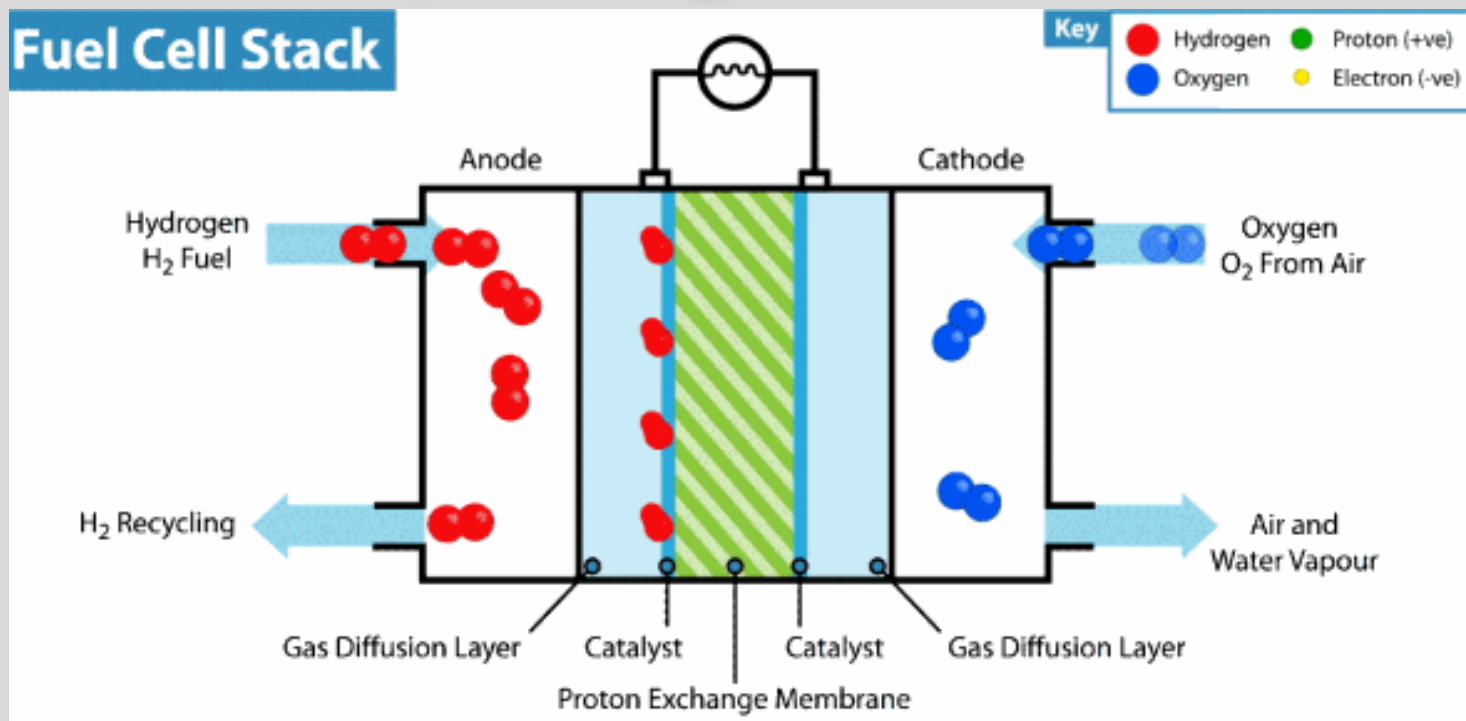
Steven Ochoa

Advisor: Dr. Yuan Li

PI: Prof. Vinayak Dravid



# The Hydrogen Solution

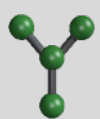


## Pros:

- Hydrogen is most abundant element in the universe
- High energy content (rocket fuel)
- No pollutants produced (only byproducts are Heat +  $H_2O$ )

## Cons:

Non-competitive due to inefficient production

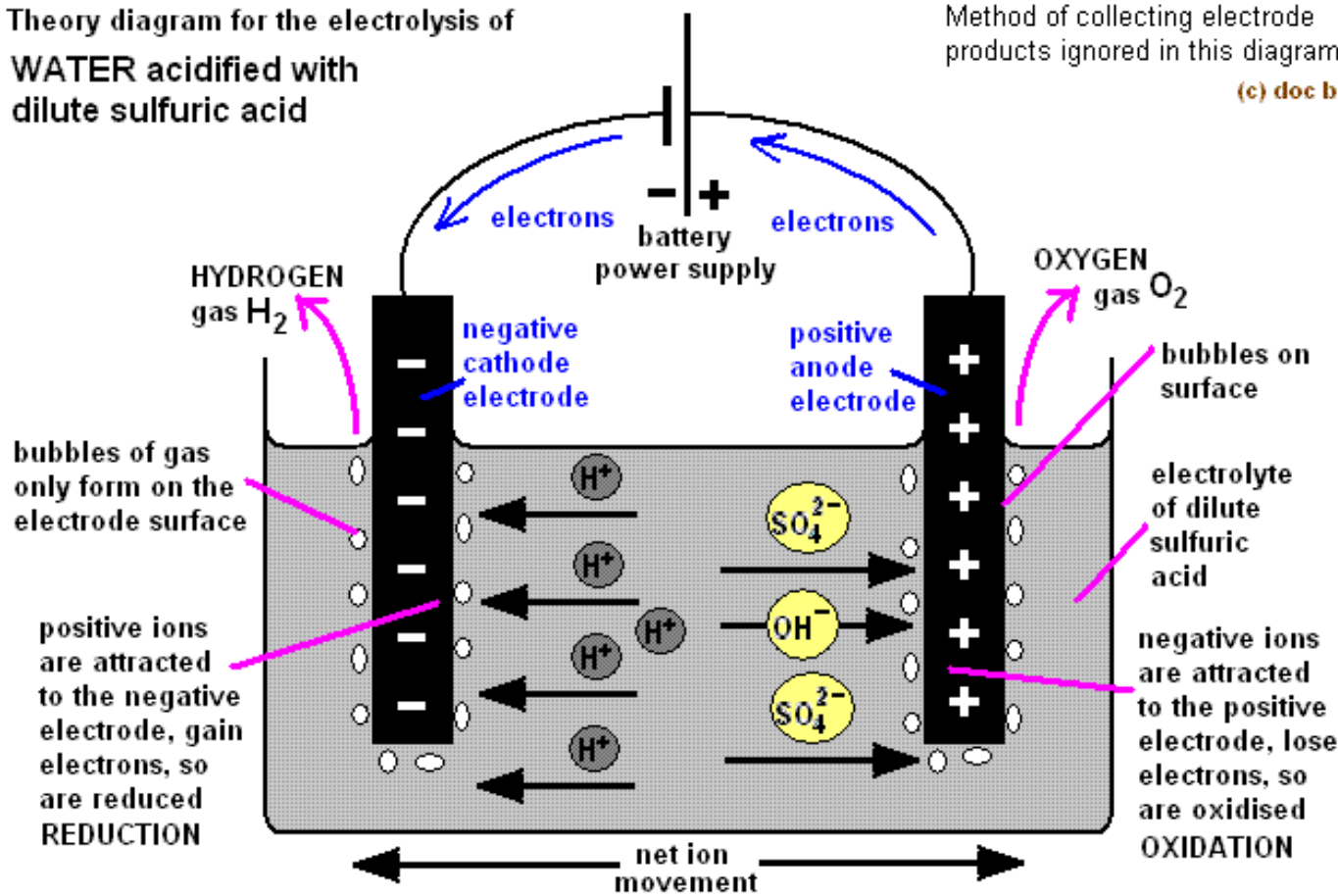


# Electrolysis

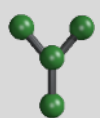
Theory diagram for the electrolysis of  
**WATER acidified with  
dilute sulfuric acid**

Method of collecting electrode  
products ignored in this diagram

(c) doc b



- Hydrogen evolution reaction occurs by means of splitting an acid into its components
- Requires a great deal of energy to operate reaction



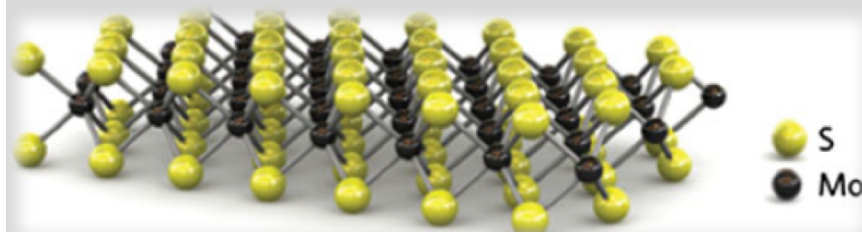
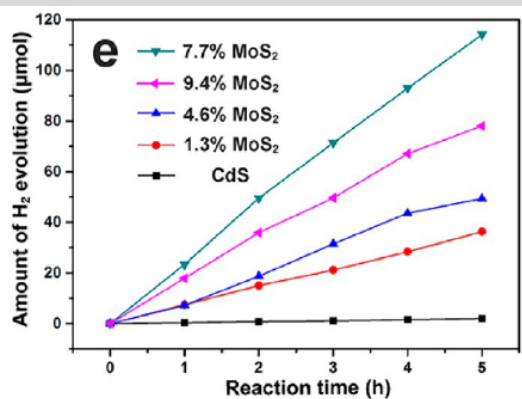
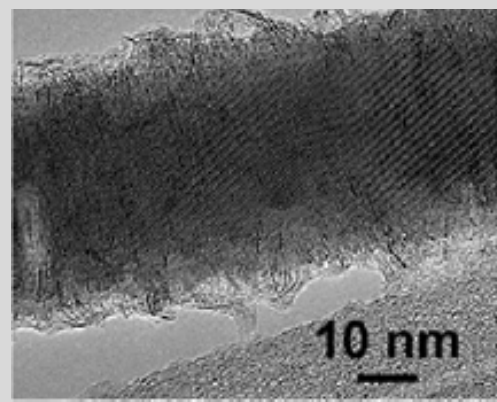
# TMDs as Catalysts for the Production of $H_2$

- Transition Metal Dichalcogenides (TMDs) are a family of layered semiconductors
  - Excellent electronic, optical, and mechanical properties
- $MoS_2$  has good stability and high catalytic activity along edges
- Has been shown to be effective in photocatalytic HER

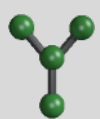
## Transition Metal Dichalcogenides - $MX_2$

Legend:

- hydrogen (black)
- alkali metals (yellow)
- alkali earth metals (orange)
- transition metals (pink)
- poor metals (light green)
- nonmetals (blue)
- noble gases (purple)
- rare earth metals (teal)

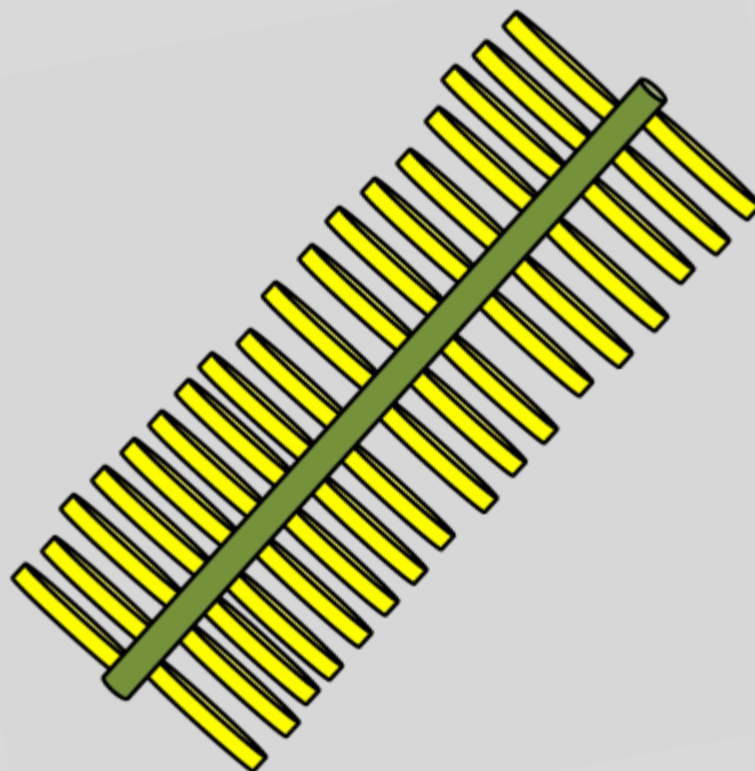


Chen, Junze, et al. Journal of the American Chemical Society (2017).



# Project Goal

- Use a morphological approach to create a heterostructure with high  $\text{MoS}_2$  edge concentration to promote efficient HER
- Nanowire base structure with flake/wing-like protrusions
- Synthesize by facile vapor deposition process



- CuS Nanowire



-  $\text{MoS}_2$  Nanoflake

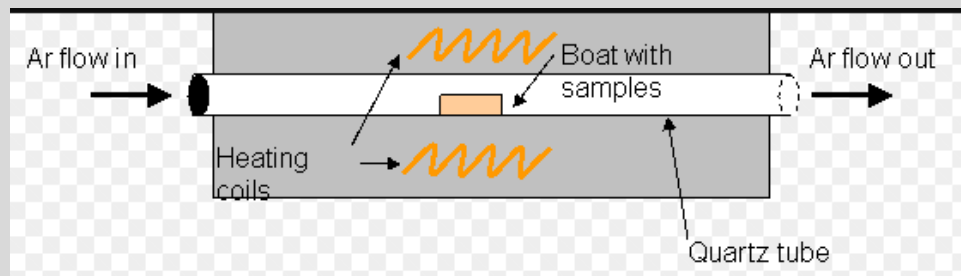
# Synthesis of $\text{Cu}_x\text{O}$ Nanowire Template

## Sample Prep

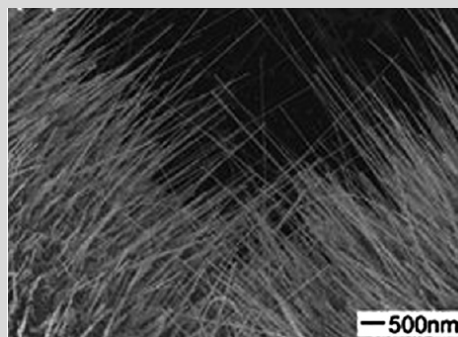
- Sonicate in Acetone - 5 minutes
- Sonicate in Isopropyl Alcohol - 5 minutes
- Sonicate in DI water - 5 minutes
- Submerge in  $\text{HNO}_3$  - 30 seconds
- Dry in  $\text{N}_2$



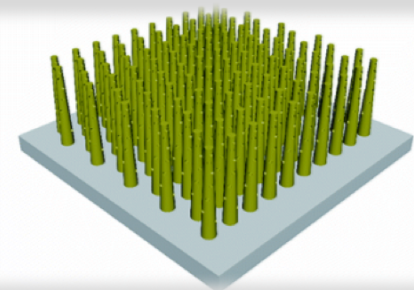
Begin with thin copper sheet



Thermal Oxidation ( $\text{N}_2$  environment)

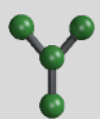


Zhang, Q et. al, (2014). *Progress in Materials Science*, 60, 208-337.



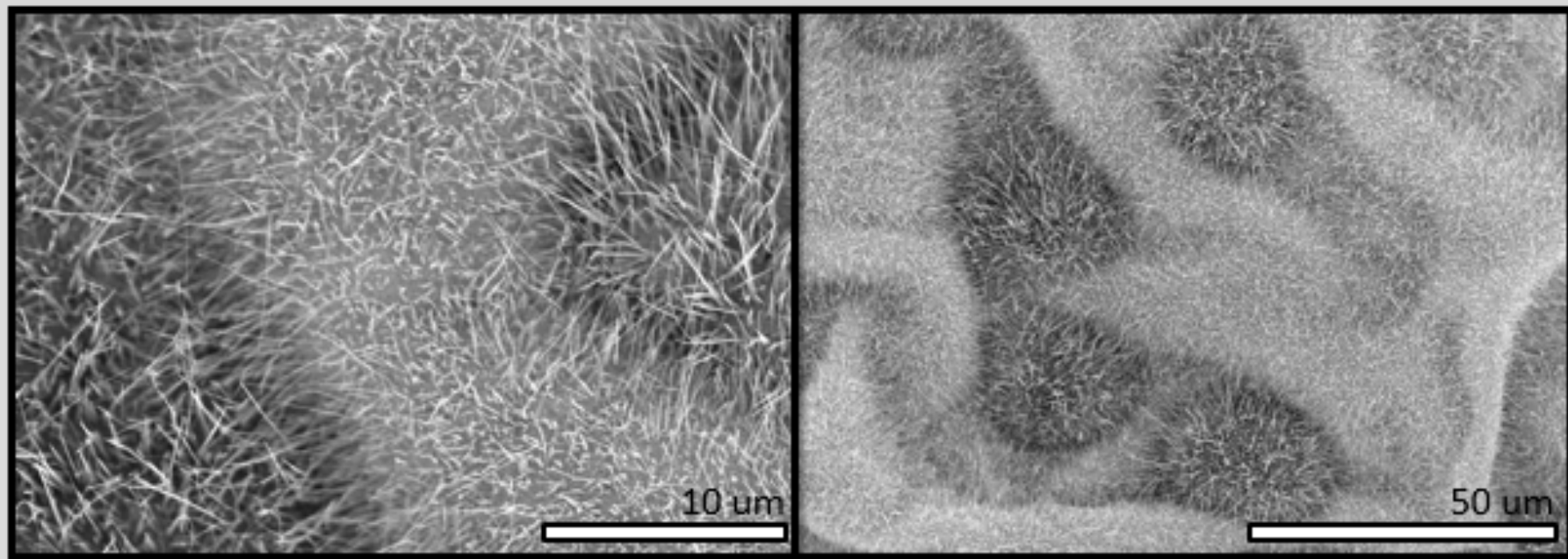
$\text{CuO/Cu}_2\text{O}$  nanowires

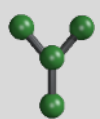




# $\text{Cu}_x\text{O}$ Nanowires

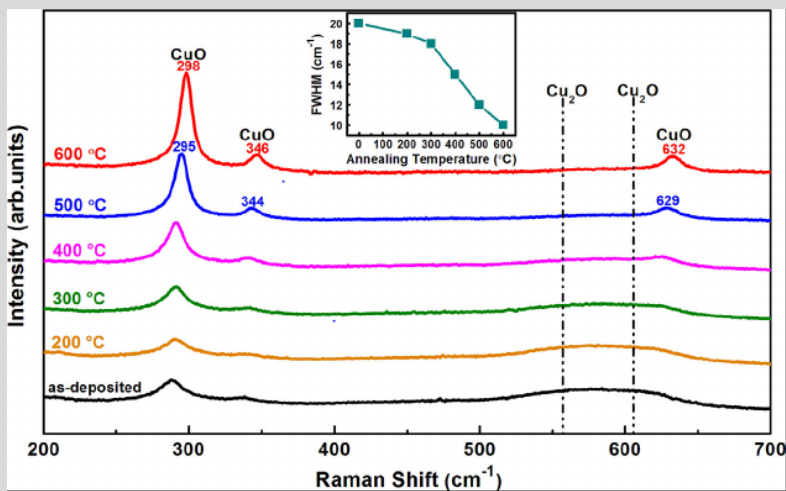
- Thermal oxidation of pure Cu substrate at 410°C for 5 hrs



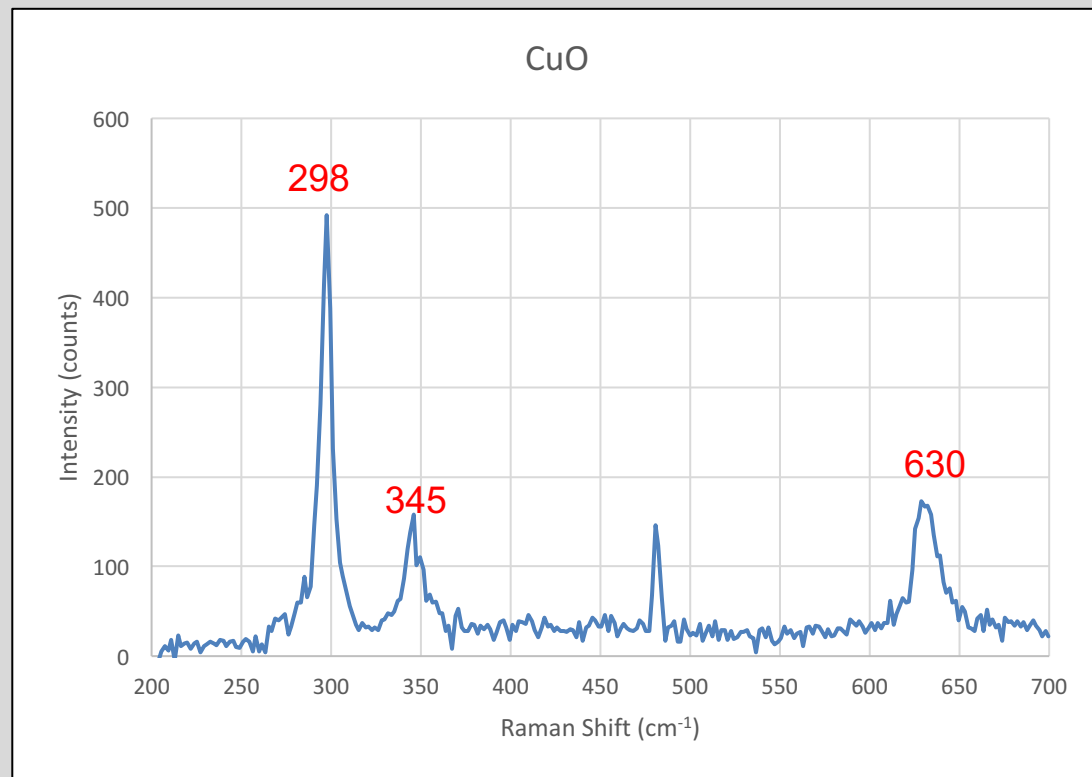


# CuO Nanowires

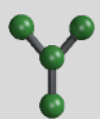
- Raman spectroscopy can measure the vibrational modes in a material which can be used to identify materials
- Spectra obtained compares well with reference
  - Same Raman stretching/bending modes detected



Akgul, Funda Aksoy, et al. "Influence of thermal annealing on microstructural, morphological, optical properties and surface electronic structure of copper oxide thin films." *Materials Chemistry and Physics*

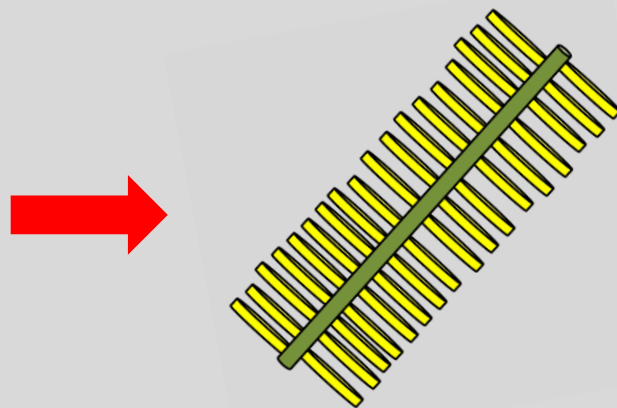
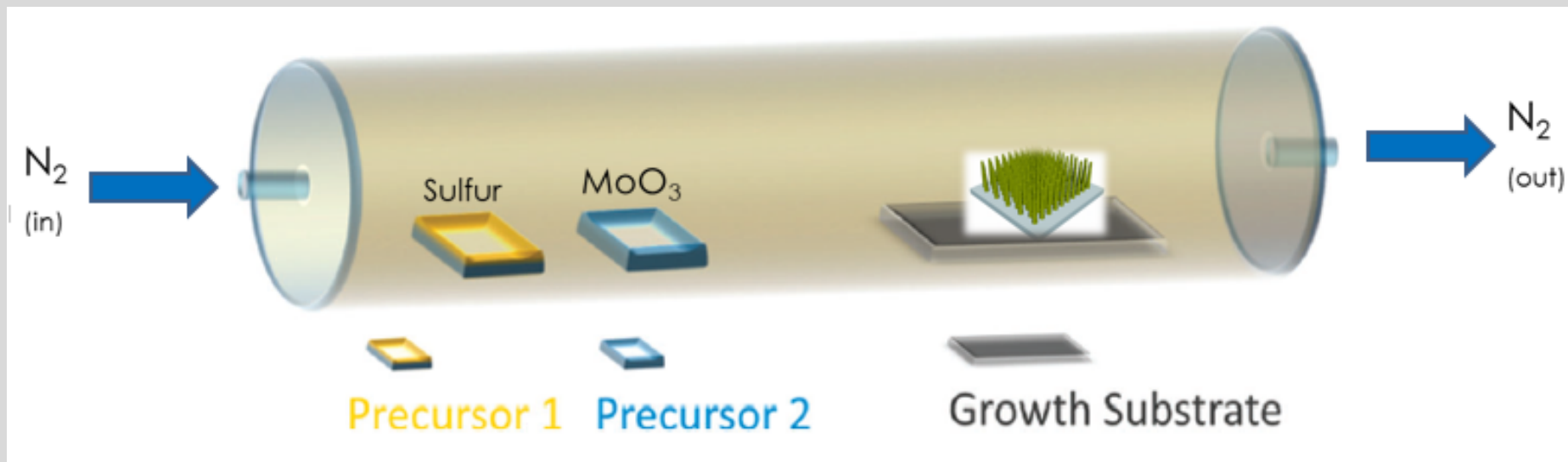


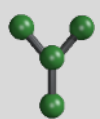




# Heterostructure Synthesis

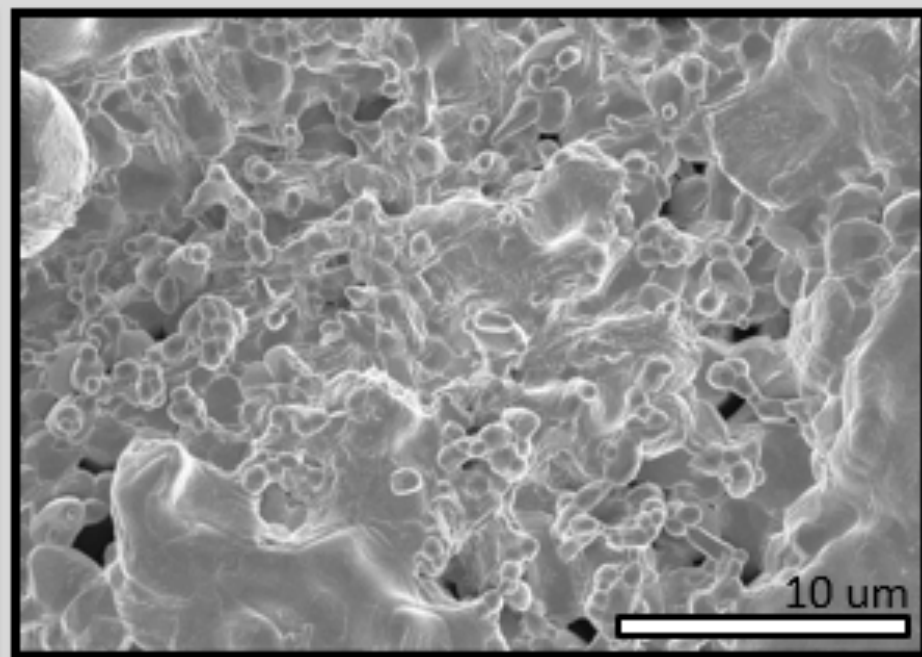
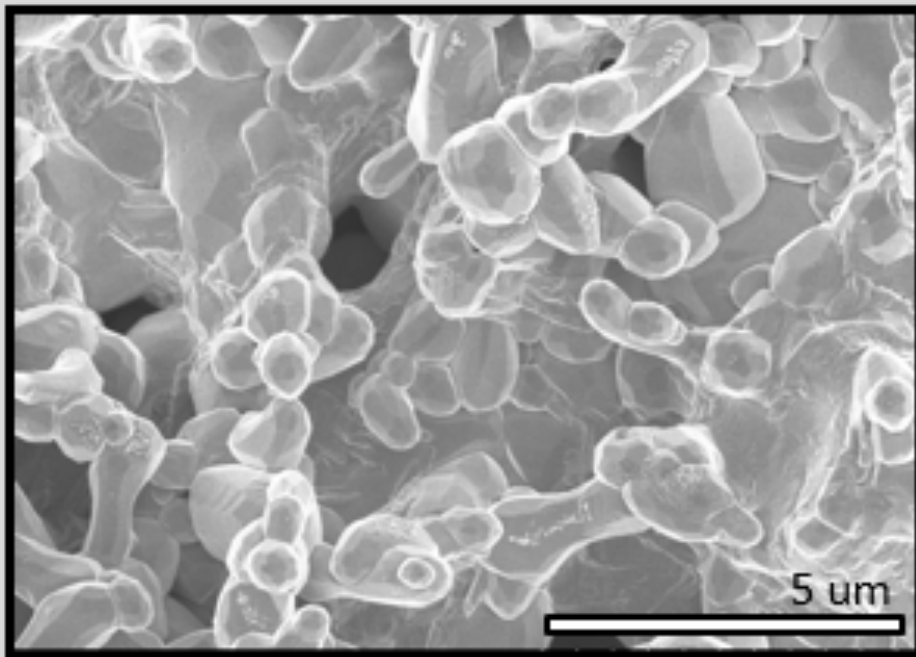
Chemical Vapor Deposition(CVD)

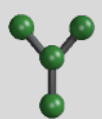




# Initial MoS<sub>2</sub> Growth Attempts

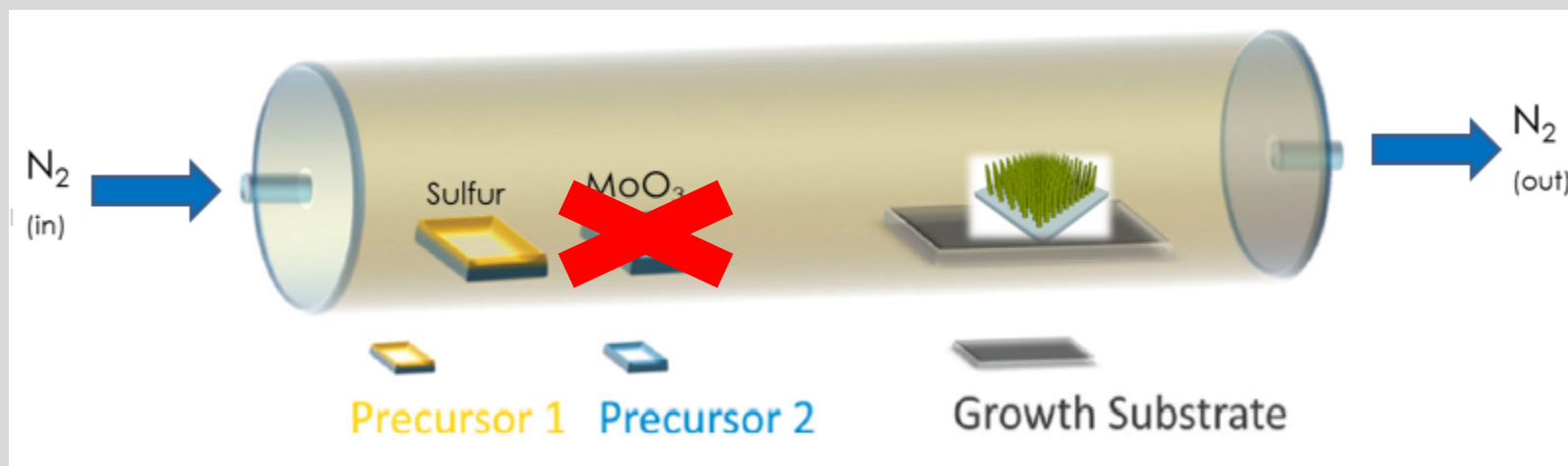
- Attempted standard MoS<sub>2</sub> growth w/ CuO nanowires at 650°C for 30 mins w/ MoO<sub>3</sub> and S precursors
- Unwanted globular structures appeared

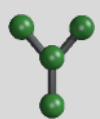




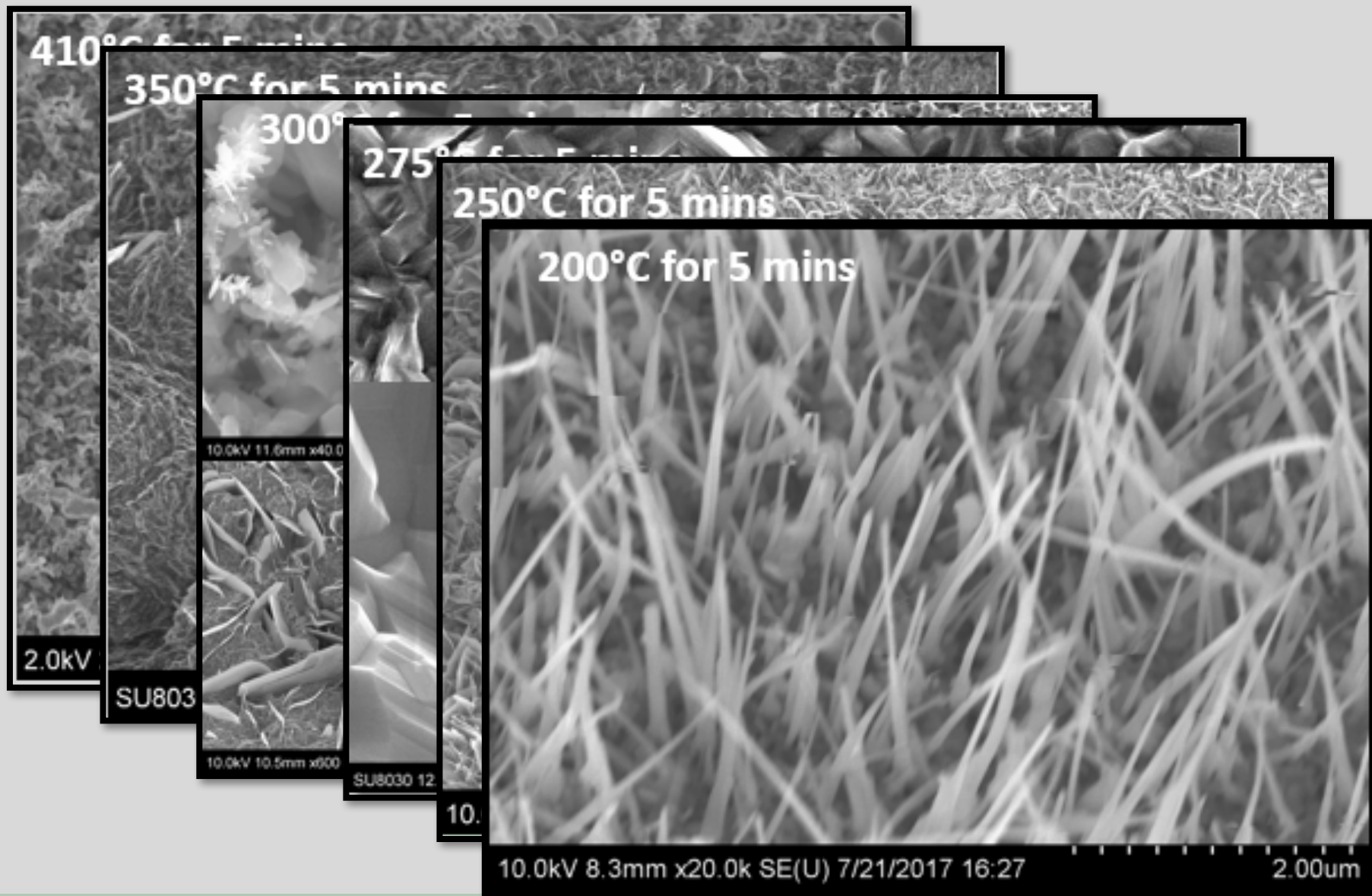
# Sulfurization Experiments

- To test the nanowire stability, the CuO nanowires were subjected to sulfurization tests at varying temperatures
- Only sulfur upstream (no  $\text{MoO}_3$ )

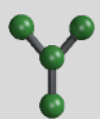




# Sulfurization Experiments







# Heterostructure Synthesis

- Deposited 1 nm thick layer of Mo on to  $\text{Cu}_x\text{O}$  nanowire surface using electron beam evaporator
- Sulfurized sample at  $300^\circ\text{C}$  for 5 mins
- Synthesis of rod-like structures with fins parallel to rod-axis (high edge concentration)



10.0kV 7.8mm x1.20k SE(U) 7/21/2017 16:15

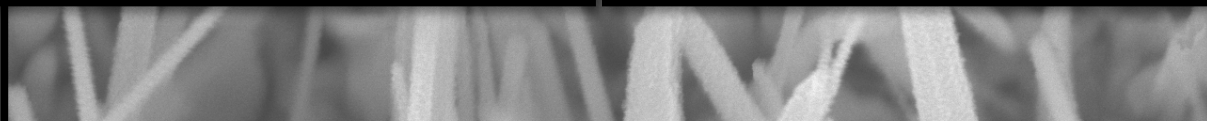
40.0um



“nano-churros”

10.0kV 7.8mm x7.00k SE(U) 7/21/2017 16:14

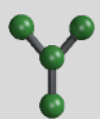
5.00um



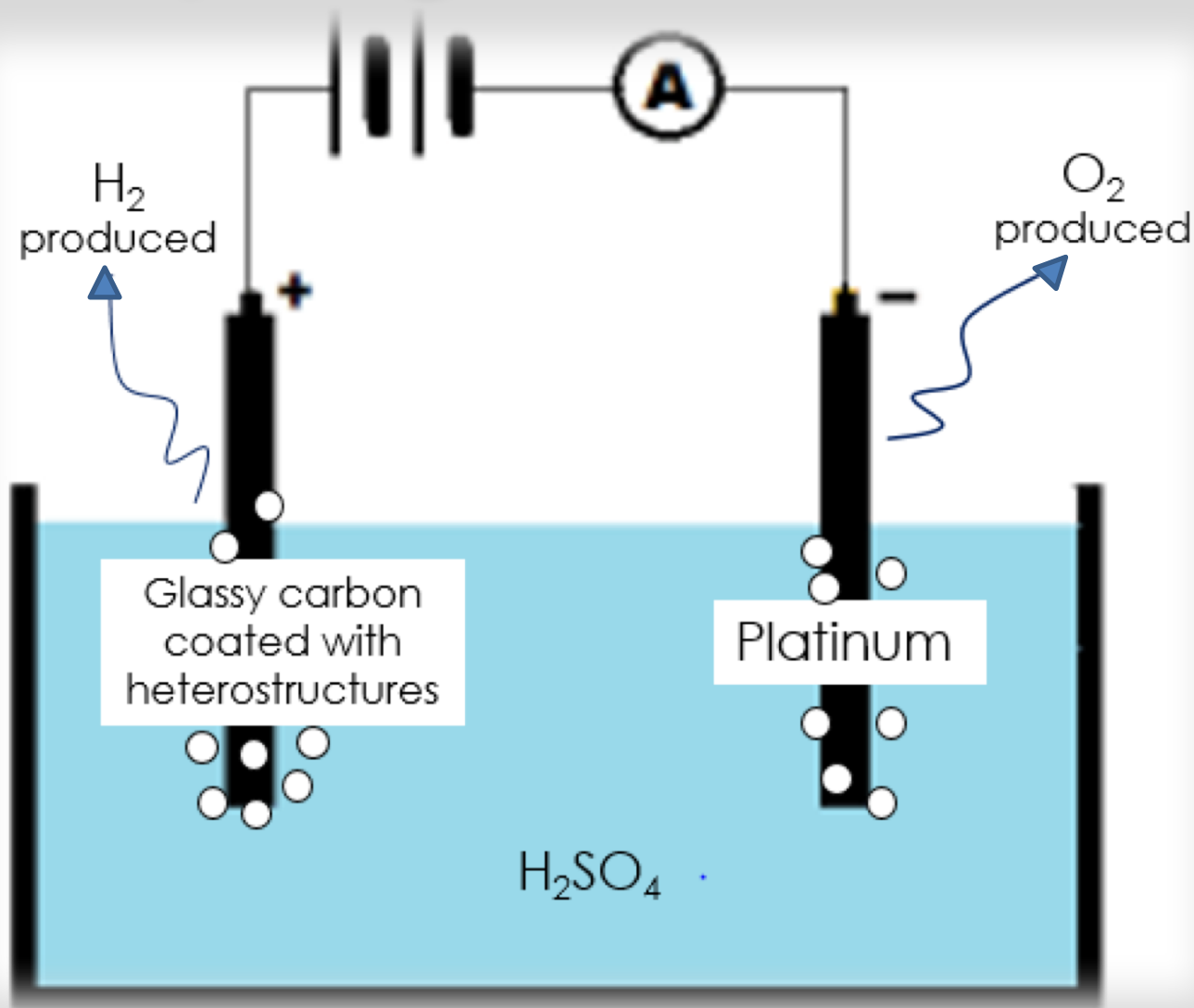
10.0kV 9.0mm x45.0k SE(U) 7/21/2017 16:20

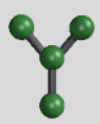
1.00um





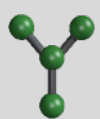
# Hydrogen Evolution





# Future Work

- Coat more samples of CuO nanowires with nanometer thick layer of Mo using electron beam evaporator
- Understand link between growth parameters and structural morphology
  - Try to recreate the “nano-churro” structures achieved before in greater quantity
- Further analyze samples using Raman, XPS, and XRD
- Measure HER using three - electrode electrochemical system test
  - Cyclic voltammetry test (CV)
  - Electrochemical Impedance Spectroscopy (EIS)
  - Generate Tafel slope to show overpotential



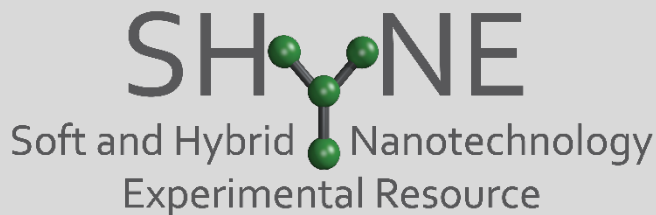
# Acknowledgements

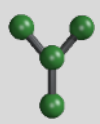
Prof. Vinayak Dravid

Dr. Yuan Li

2-D Subgroup: Akshay Murthy, Eve Hanson, Jennifer DiStefano

This work made use of the EPIC, Keck-II, and/or SPID facility(ies) of Northwestern University's NUANCE Center, which has received support from the Soft and Hybrid Nanotechnology Experimental (SHyNE) Resource (NSF ECCS-1542205); the MRSEC program (NSF DMR-1121262) at the Materials Research Center; the International Institute for Nanotechnology (IIN); the Keck Foundation; and the State of Illinois, through the IIN.





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