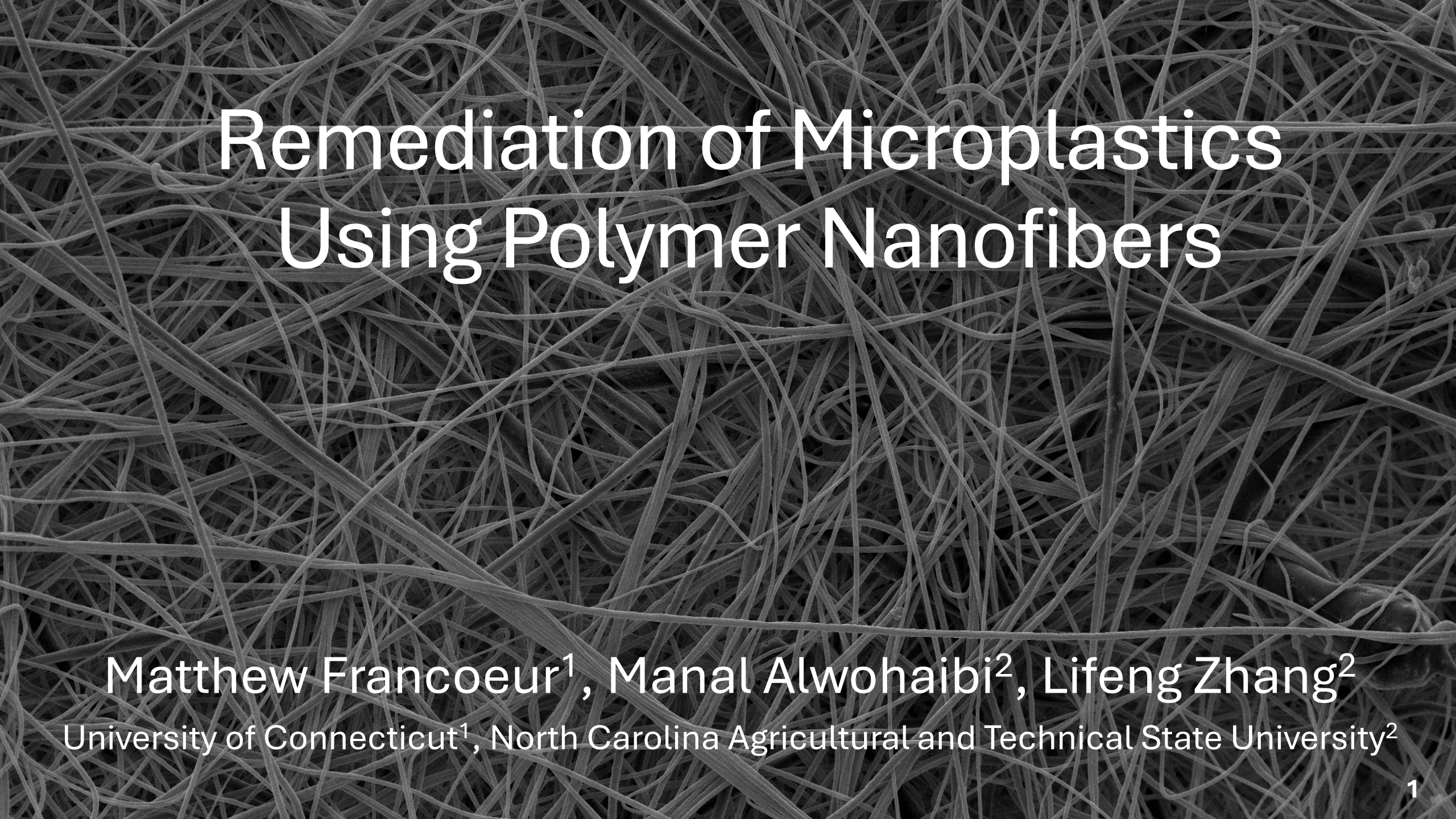




Remediation of Microplastics Using Polymer Nanofibers

Matthew Francoeur¹, Manal Alwohaibi², Lifeng Zhang²

University of Connecticut¹, North Carolina Agricultural and Technical State University²

Microplastic (MP) Pollution

- **The average person consumes 11,845 and 193,200 microplastic (MP) particles per year**

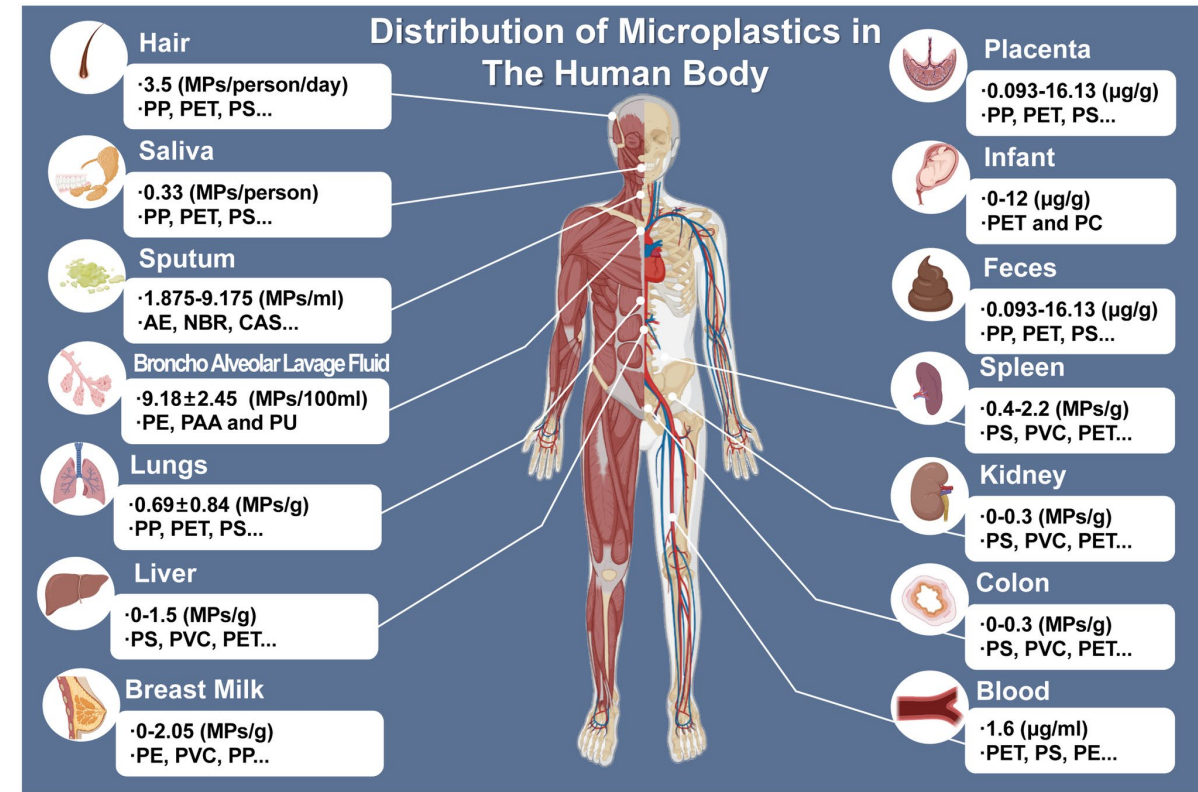
(Hoang et al., *Frontiers in Environmental Science*, June 2025)

- **Drinking water is the primary route of MP exposure**

(Hoang et al., *Frontiers in Environmental Science*, June 2025)

- **Current methods of removal in water treatment often struggle with smaller particles ($<50\mu\text{m}$)**

(Shahariar et al., *Journal of Water Process Engineering*, May 2026)



(Yang et al., *Current Pollution Reports*, August 2023)

Current Remediation Methods

Flocculation/Sedimentation

- Use of chemical agents to clump and sediment small particles.
- <0.1% removal of 1 μ m MP particles.

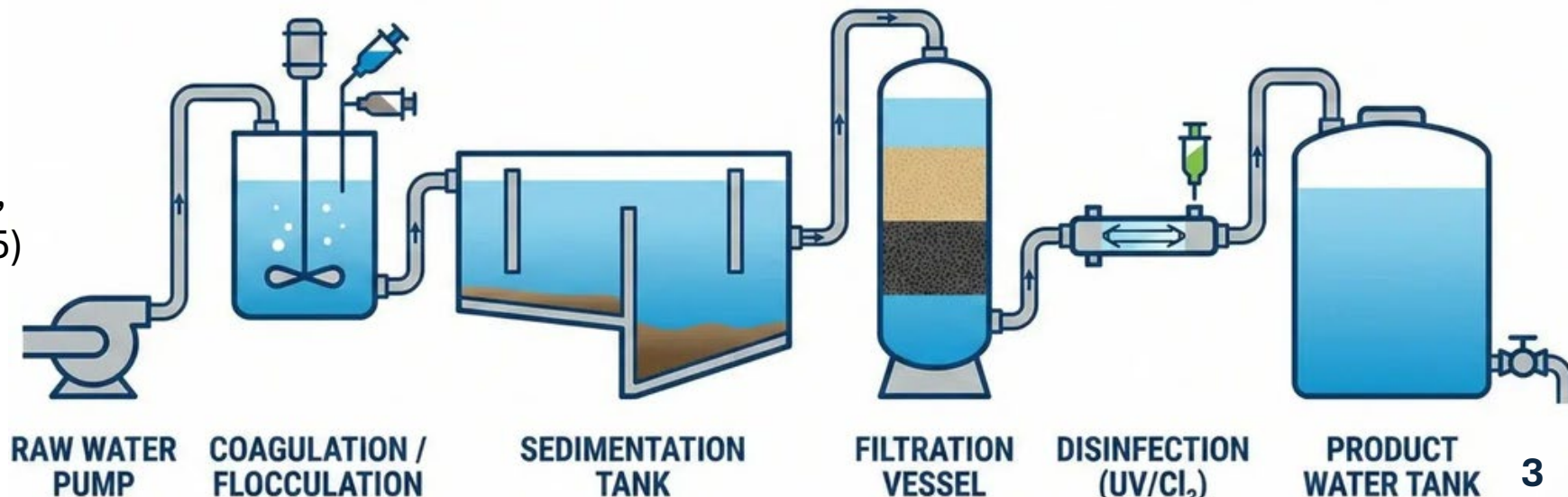
(Shahariar et al., *Journal of Water Process Engineering*, May 2026)

Granular Filtration

- Uses sand or activated carbon in large (**~50cm tall**) filter columns
- Completely removes microplastics >100 μ m
- Microplastics <50 μ m have worse removal efficiency (85-95%)

(Zhang et al., *Science of The Total Environment*, June 2020)

(COVNA STARK,
December 2025)



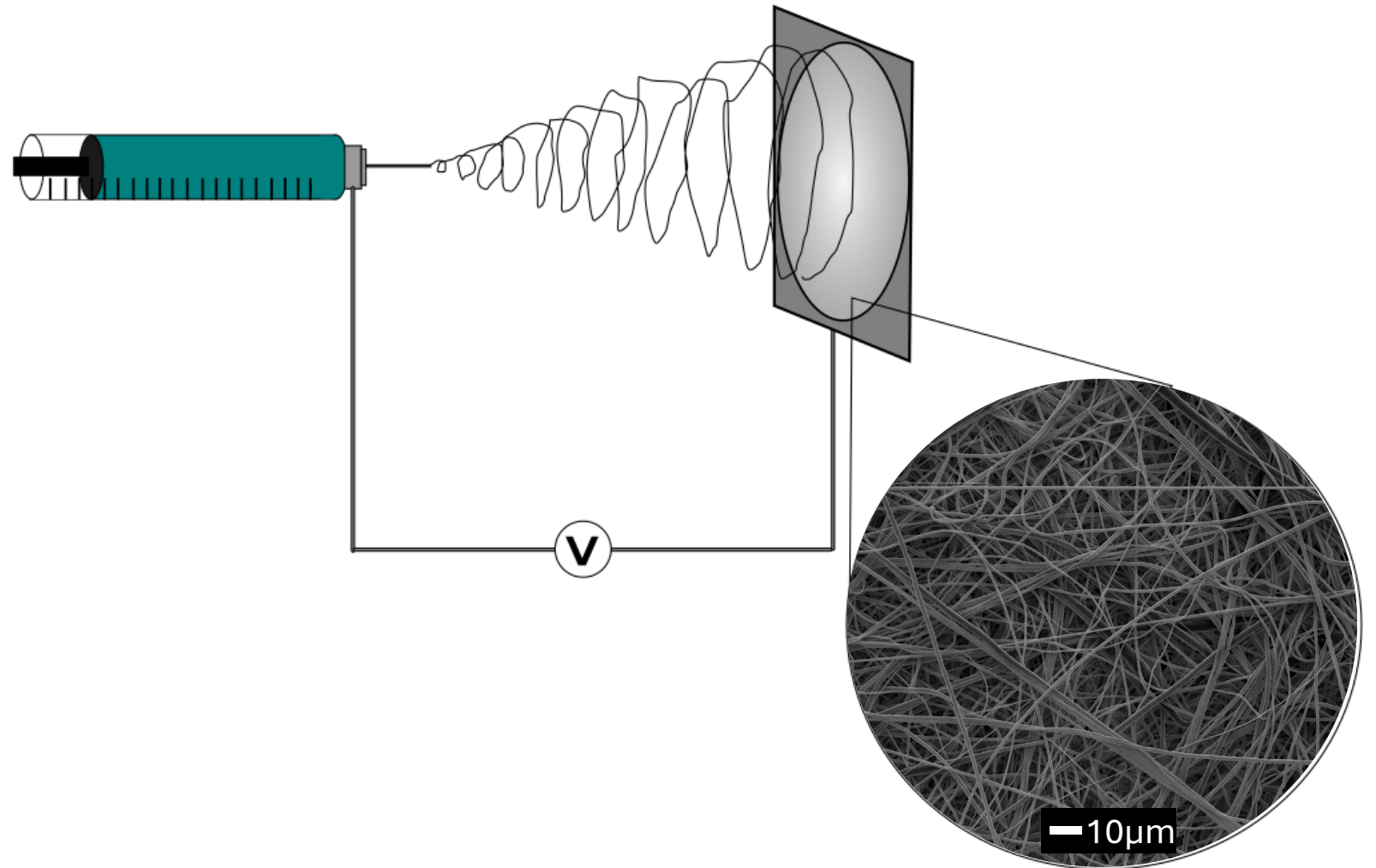
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Nanofibers & Electrospinning

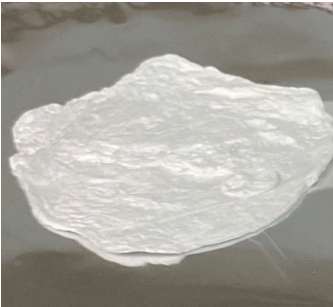
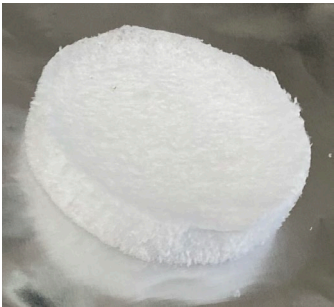
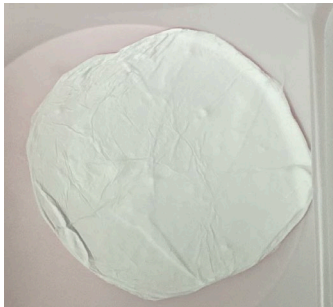
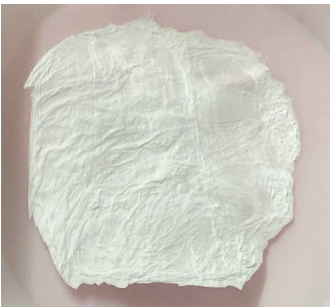
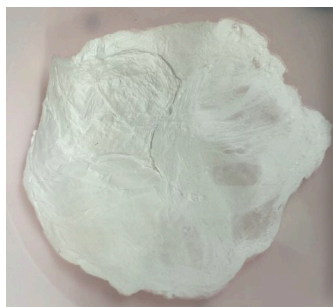
- High voltage draws polymer solution to nm sizes.
- High surface area for adsorption.
- Small pore size for size exclusion.



Survey of Nanomaterial Filters

What nanomaterial filters work best for MP removal?

- **What chemistry is best?** : Cellulose Acetate, Cellulose, Chitosan, Polyacrylonitrile (PAN)
- **What morphology is best?** : Membrane or Aerogel

PAN Nanofiber Membrane	PAN Nanofiber Aerogel	Cellulose Acetate Nanofiber Membrane	Cellulose Nanofiber Membrane	Cellulose Nanofiber Aerogel	Cellulose Nanofiber & Chitosan Membrane
					



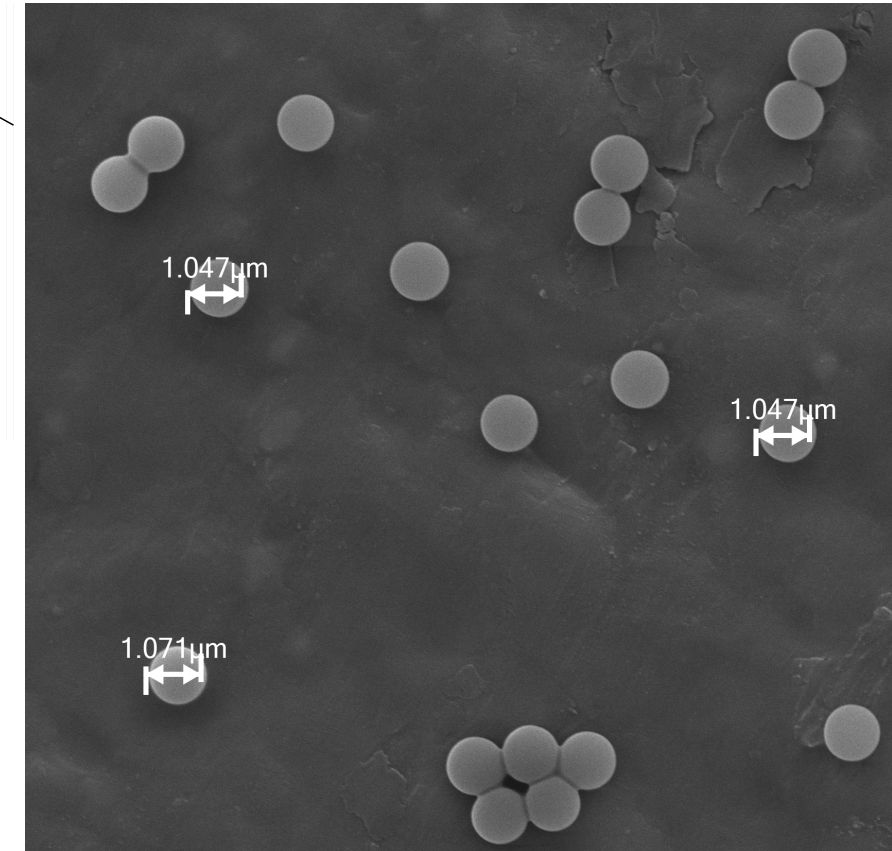
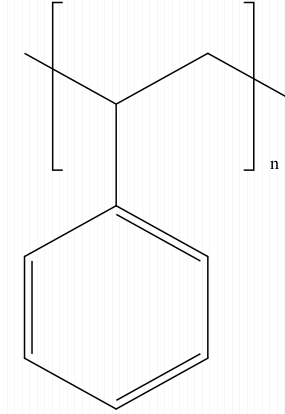
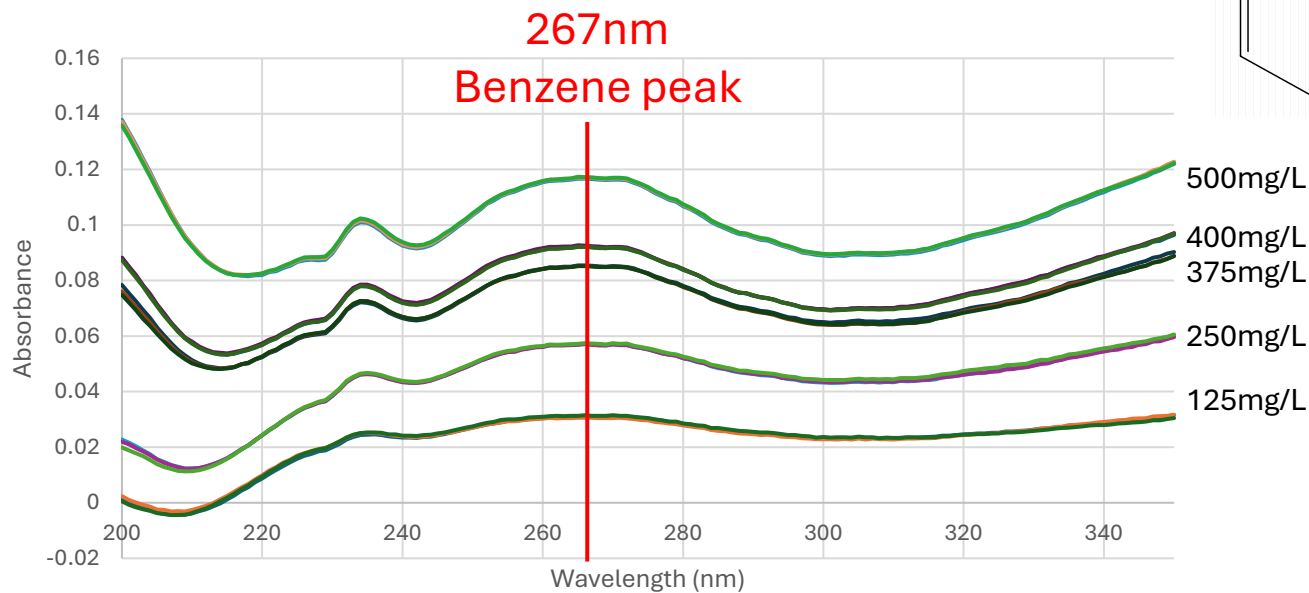
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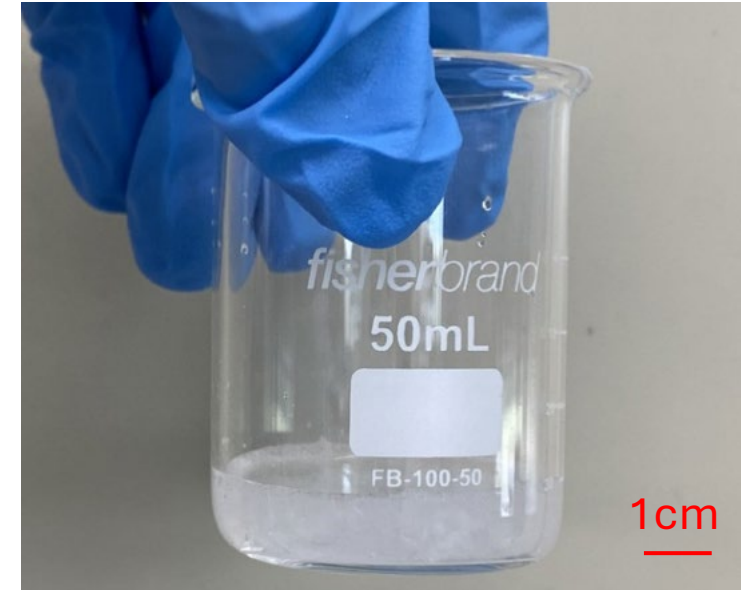
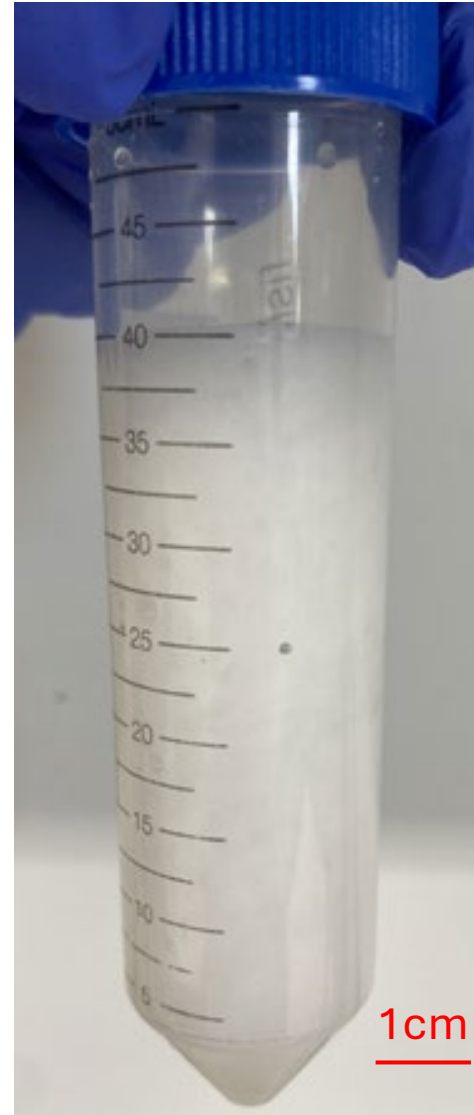
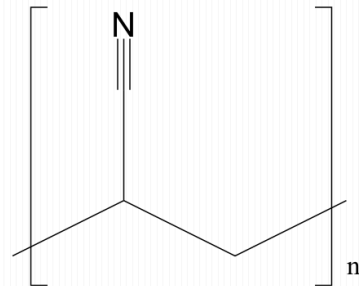
Microplastic Pollution Model

- 1 μ m Polystyrene Spheres
- Quantified by UV-Vis



Polyacrylonitrile (PAN) Fibers

- Electrospun in Dimethylformamide (DMF)
(Zhang, *Journal of Membrane Science*, March 2011)
- Used as membrane and aerogel
- Aerogel formed by ultrasonication → freeze drying



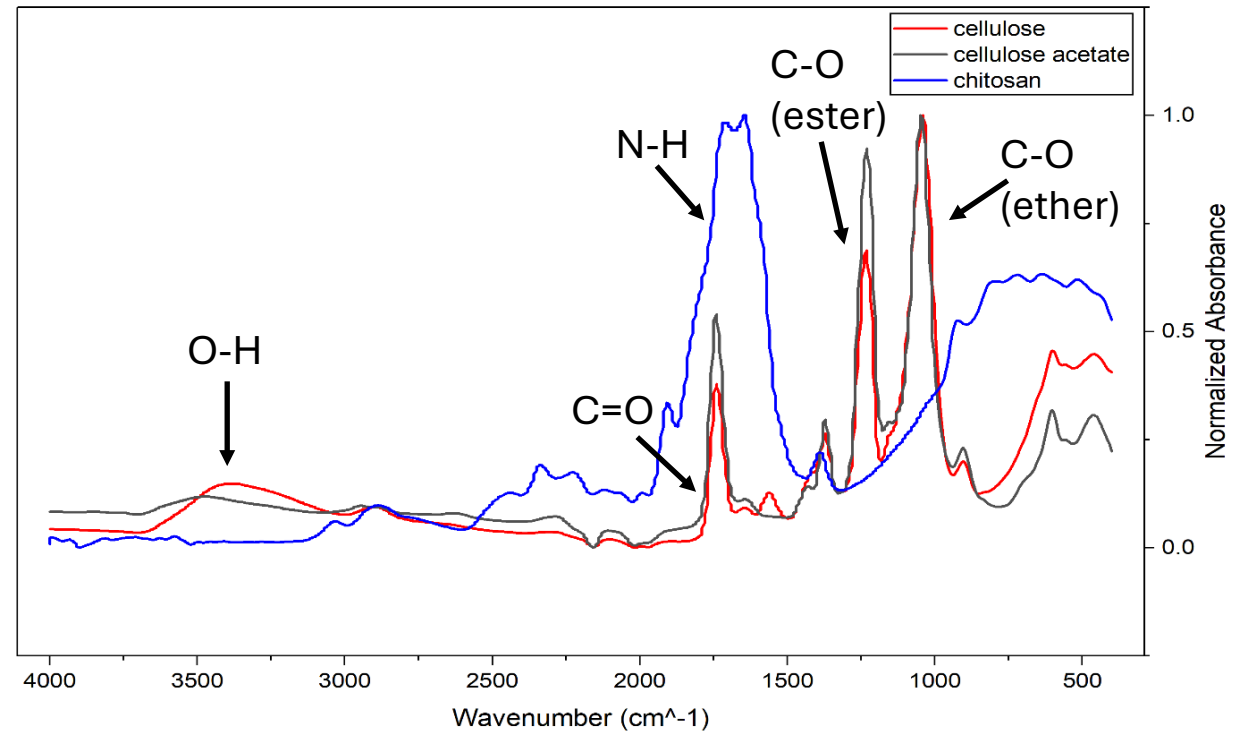
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Cellulose Fibers

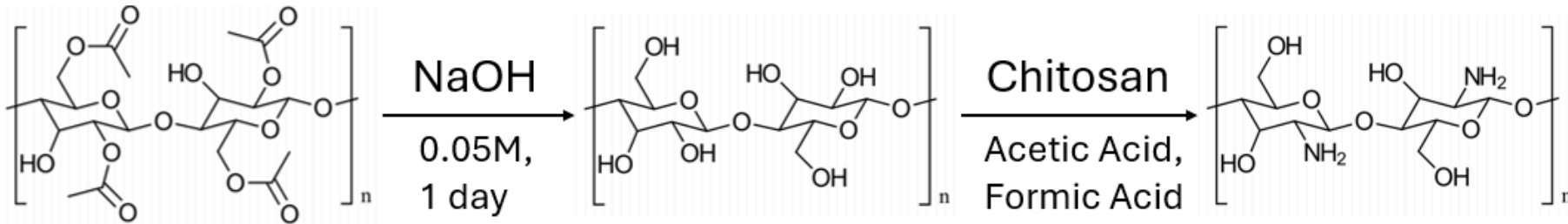
- Cellulose Acetate is directly Electrospun in Acetone and DMSO (Dimethyl Sulfoxide)
- Other cellulose-based materials are formed from Cellulose Acetate Fiber
(Mantripragada, *Cellulose*, June 2023)



Cellulose Acetate

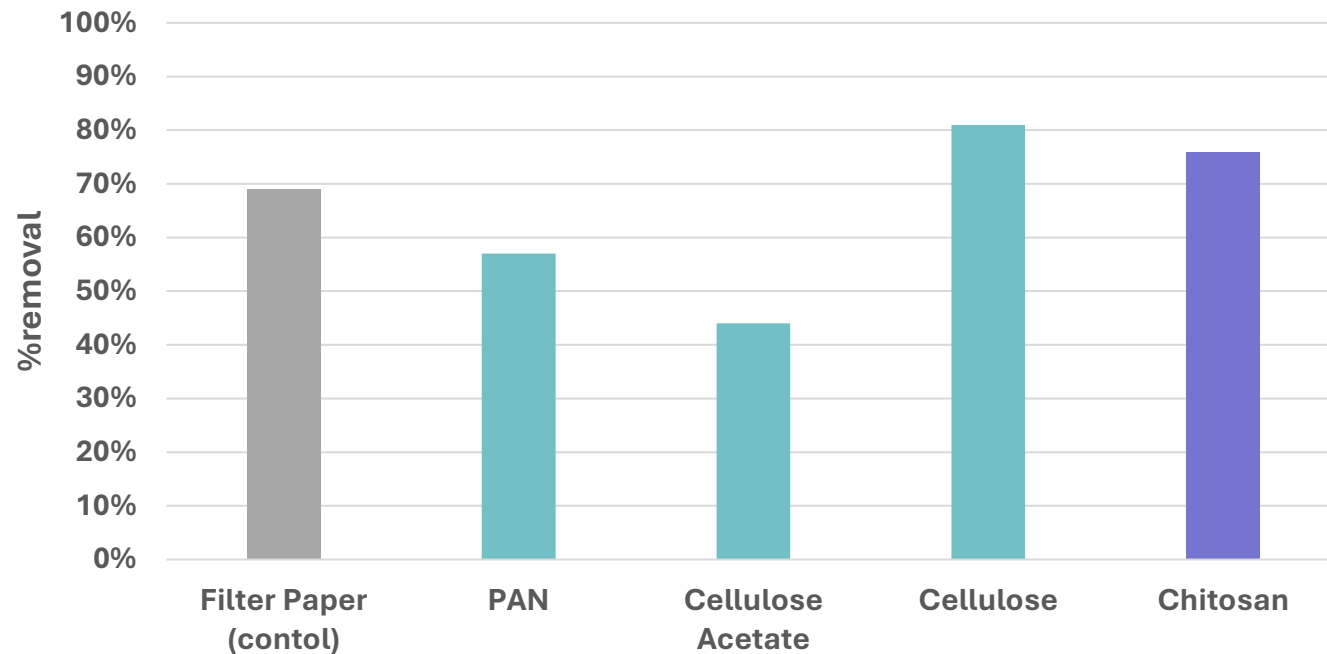
Cellulose

Chitosan-coated
Cellulose

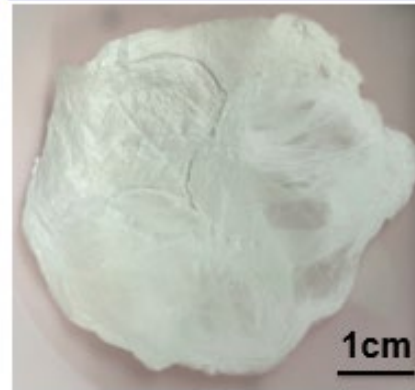


Membrane Performance

- Cellulose performs best
- All filters remove some MP



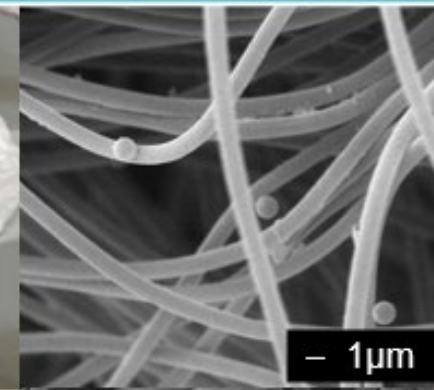
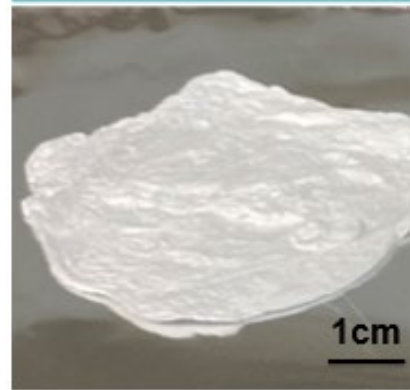
Coated Membranes



Chitosan-Dipped Cellulose

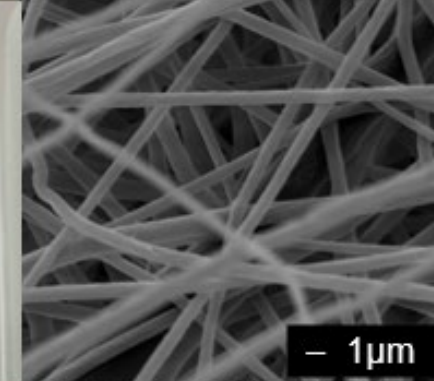
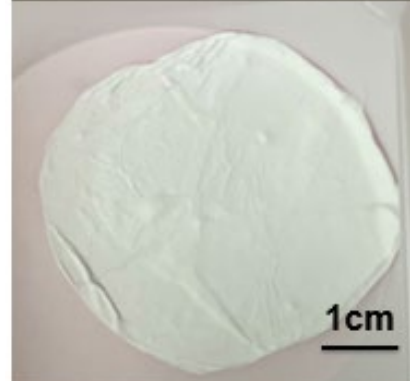
76% MP Removal

Membranes



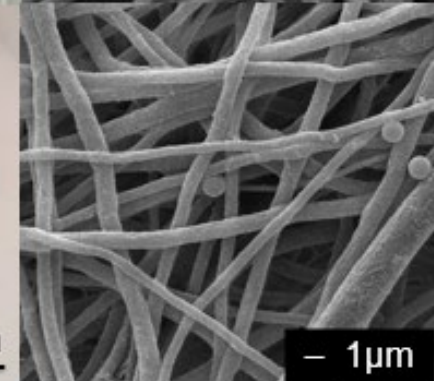
PAN
57% MP Removal

Avg: 960nm
SD: 90nm



Cellulose Acetate
44% MP Removal

Avg: 957nm
SD: 449nm

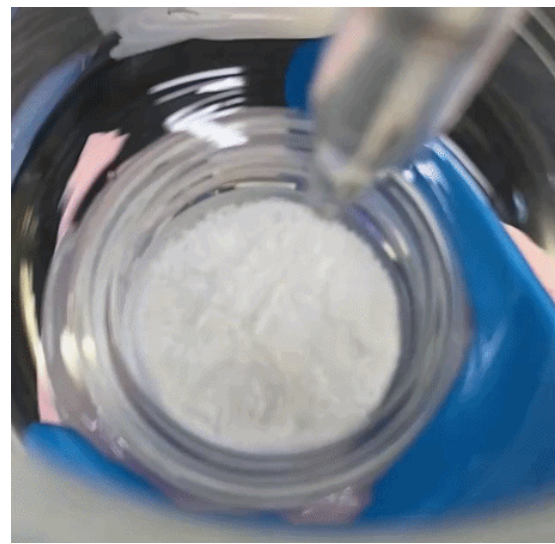
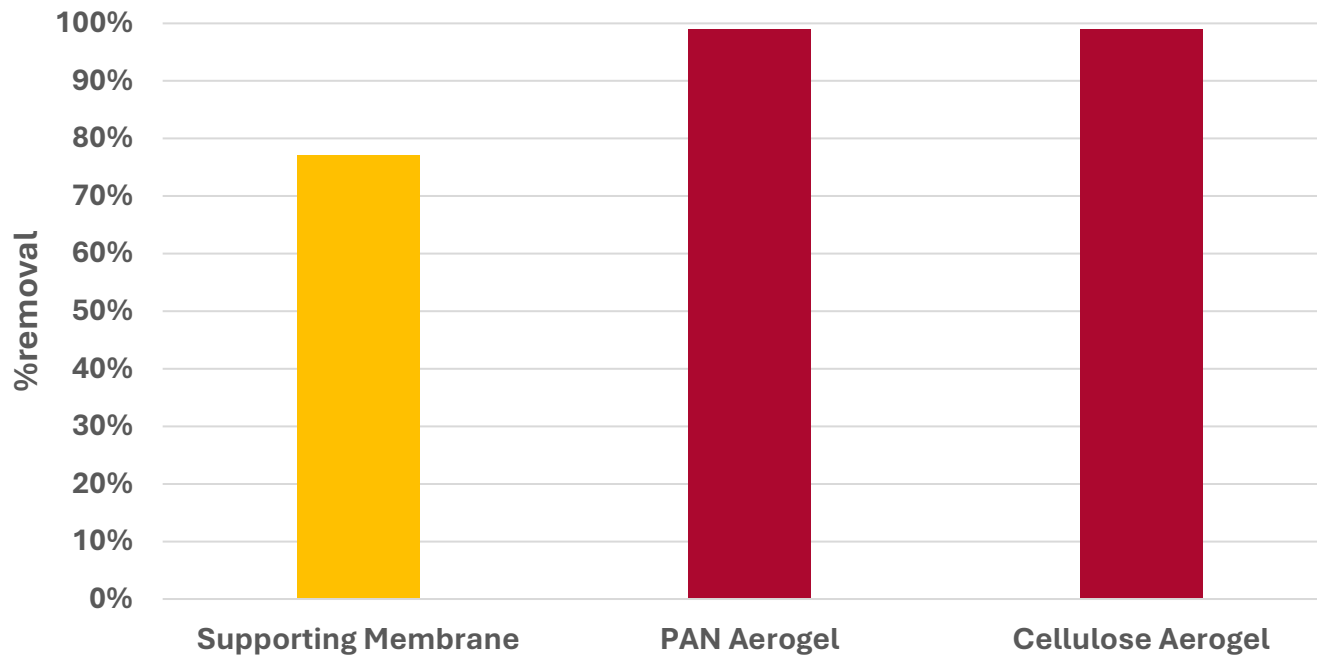


Cellulose
81% MP Removal

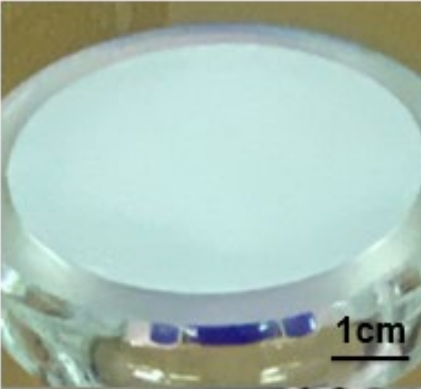
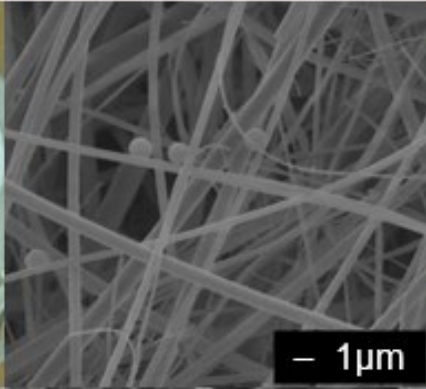
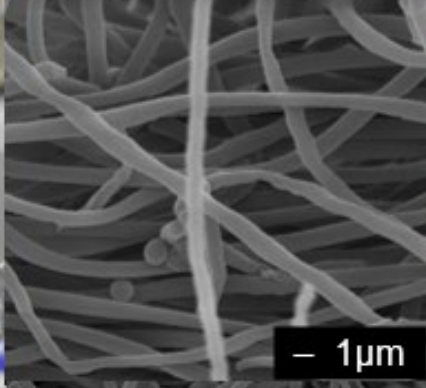
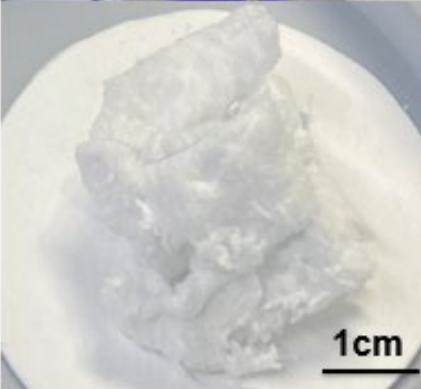
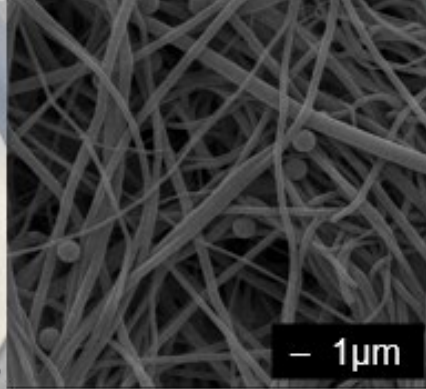
Avg: 778nm
SD: 366nm

Aerogel Performance

- Aerogels are too weak to test without support
- All aerogels remove ~100% (within resolution of UV-Vis)

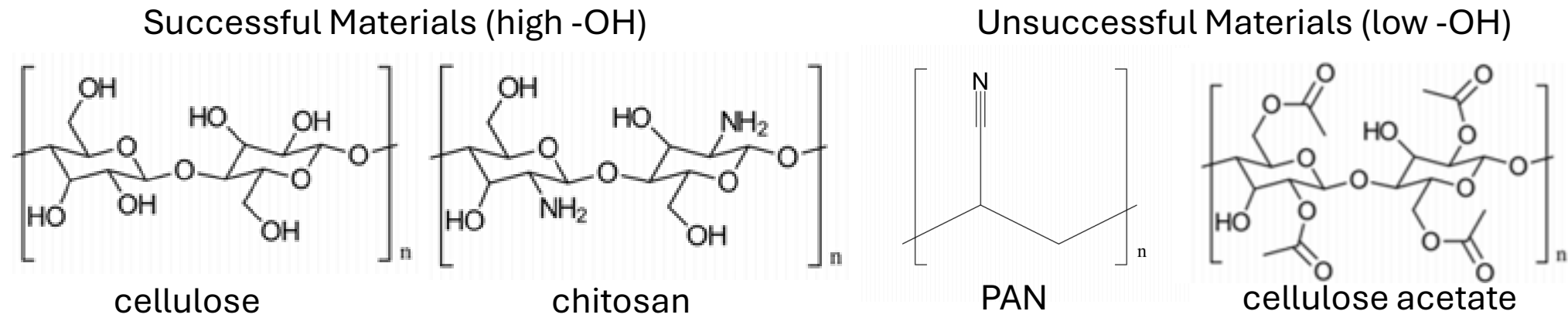


Aerogels

 1cm	 – 1μm	Supporting Membrane 77% MP Removal Avg: 673nm SD: 600nm
 1cm	 – 1μm	PAN Aerogel ~100% MP Removal Avg: 839nm SD: 89nm
 1cm	 – 1μm	Cellulose Aerogel ~100% MP Removal Avg: 412nm SD: 272nm

Conclusions

- Hydroxyl groups improve removal significantly
 - Hydrophilicity
 - Aromatic hydrogen bonding (*Schiener et al., JACS, October 2002*)
- Aerogel removes significant MP despite loose/open structure and regardless of chemistry (PAN/cellulose)
- Future Work:
 - Crosslinked Aerogels
 - MP detection at lower concentration



References

- 1) Hoang, Hong Giang, et al. “A review of microplastic pollution and human health risk assessment: Current knowledge and future outlook.” *Frontiers in Environmental Science*, vol. 13, 18 June 2025, <https://doi.org/10.3389/fenvs.2025.1606332>.
- 2) Mantripragada, Shobha, et al. “Sustainable filter/adsorbent materials from cellulose-based electrospun nanofibrous membranes with soy protein coating for high-efficiency GenX fluorocarbon remediation from water.” *Cellulose*, vol. 30, no. 11, 15 June 2023, pp. 7063–7078, <https://doi.org/10.1007/s10570-023-05304-7>.
- 3) Scheiner, Steve, et al. “Comparison of various types of hydrogen bonds involving aromatic amino acids.” *Journal of the American Chemical Society*, vol. 124, no. 44, 15 Oct. 2002, pp. 13257–13264, <https://doi.org/10.1021/ja027200q>.
- 4) Shahariar, Md Nayeem, et al. “Fate and distribution of microplastics in a small-scale drinking water treatment plant: Evidence from a year-long monitoring campaign.” *Journal of Water Process Engineering*, vol. 87, May 2026, p. 109993, <https://doi.org/10.1016/j.jwpe.2026.109993>.
- 5) *Water Treatment Process Comparison: The 4 Pillars of Purification*, COVNA STARK, 16 Dec. 2025, stark-water.com/blog/water-treatment-process-comparison/.
- 6) Yang, Ziye, et al. “Human microplastics exposure and potential health risks to target organs by different routes: A review.” *Current Pollution Reports*, vol. 9, no. 3, 17 Aug. 2023, pp. 468–485, <https://doi.org/10.1007/s40726-023-00273-8>.
- 7) Zhang, Lifeng, et al. “Antimicrobial nano-fibrous membranes developed from electrospun polyacrylonitrile nanofibers.” *Journal of Membrane Science*, vol. 369, no. 1–2, Mar. 2011, pp. 499–505, <https://doi.org/10.1016/j.memsci.2010.12.032>.
- 8) Zhang, Yongli, et al. “Removal efficiency of micro- and nanoplastics (180 nm–125 µm) during drinking water treatment.” *Science of The Total Environment*, vol. 720, June 2020, p. 137383, <https://doi.org/10.1016/j.scitotenv.2020.137383>.

Acknowledgements

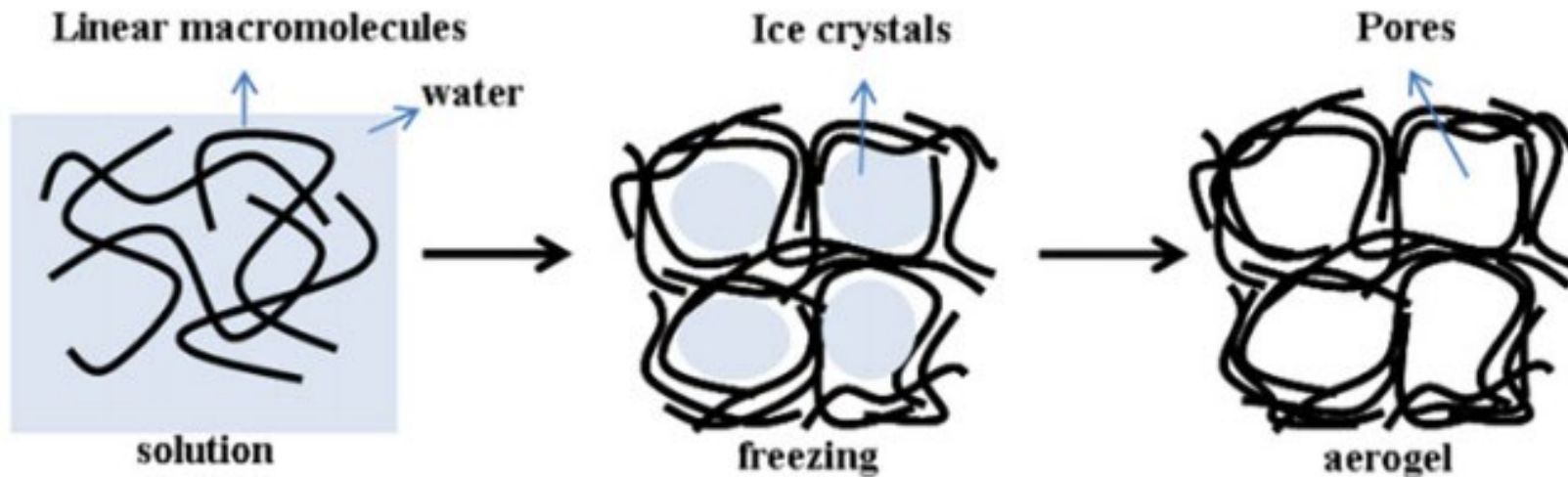
This material is based upon work supported by the National Science Foundation under award No. [#2349519] through the 2026 SENIC REU at JSNN.

Supplemental:

- Aerogel Pore Formation
- Additional SEM of Aerogels

Aerogel Formation and Freezing

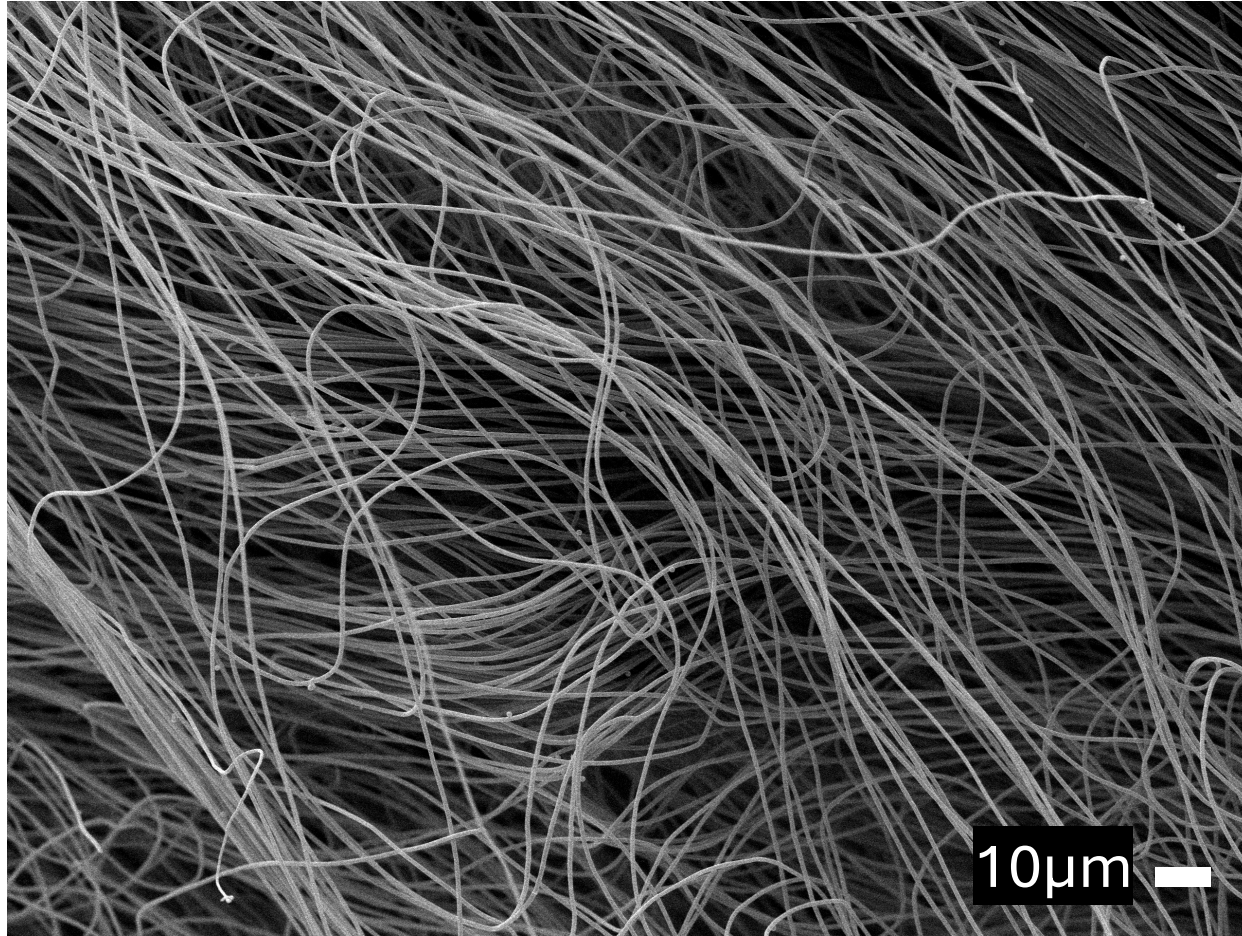
- Ice crystals open microscale pores but compact nanofibers.
- Freezing instantly in liquid nitrogen rather than freezer may create smaller ice crystals and finer porosity.



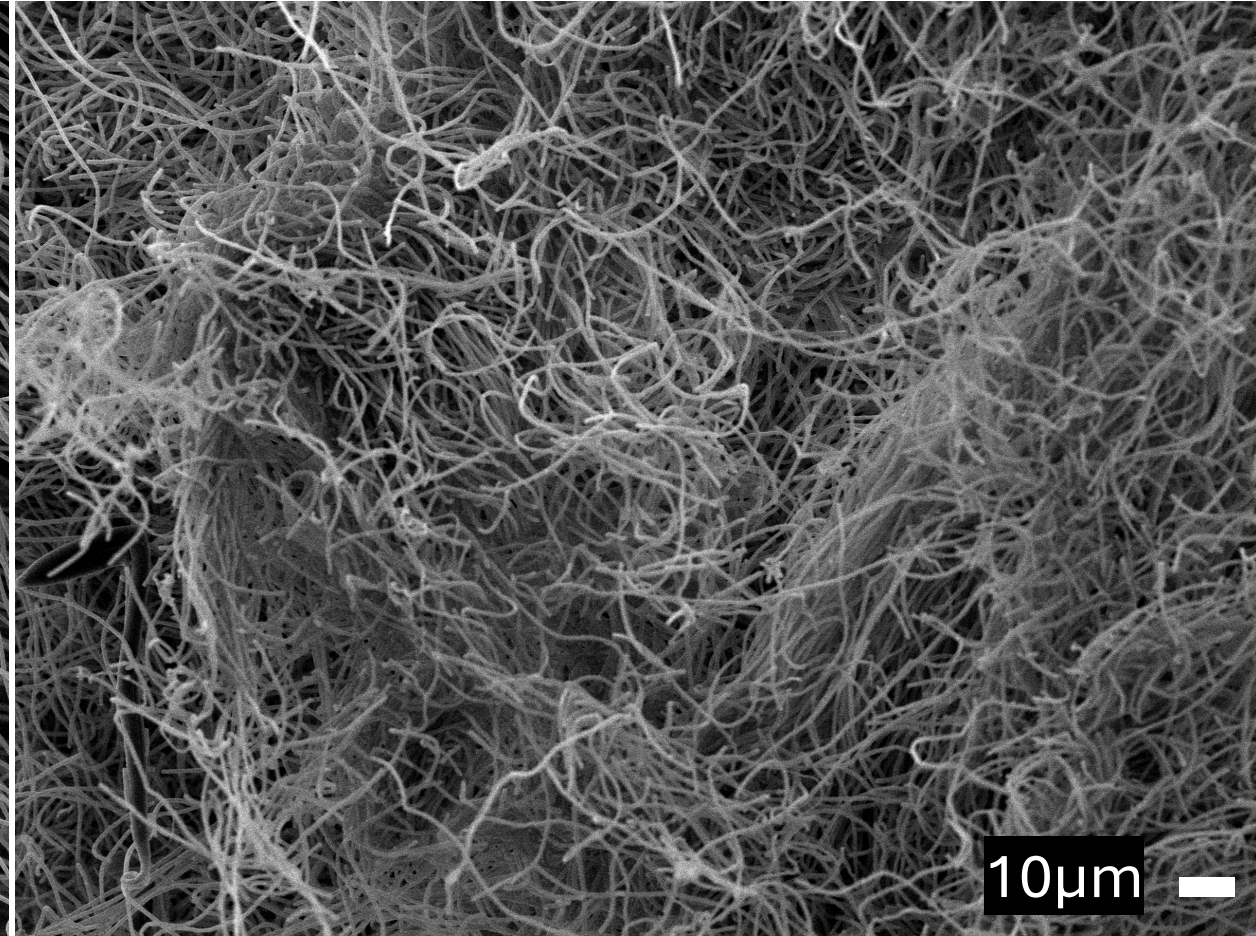
Yixin Wang, Xi Chen, Ying Kuang, Man Xiao, Yuehong Su, Fatang Jiang, Microstructure and filtration performance of konjac glucomannan-based aerogels strengthened by wheat straw, *International Journal of Low-Carbon Technologies*, Volume 14, Issue 3, September 2019, Pages 335–343, <https://doi.org/10.1093/ijlct/ctx021>

Zoomed Out Aerogel Pictures

- Morphology differences easier to see at lower magnification



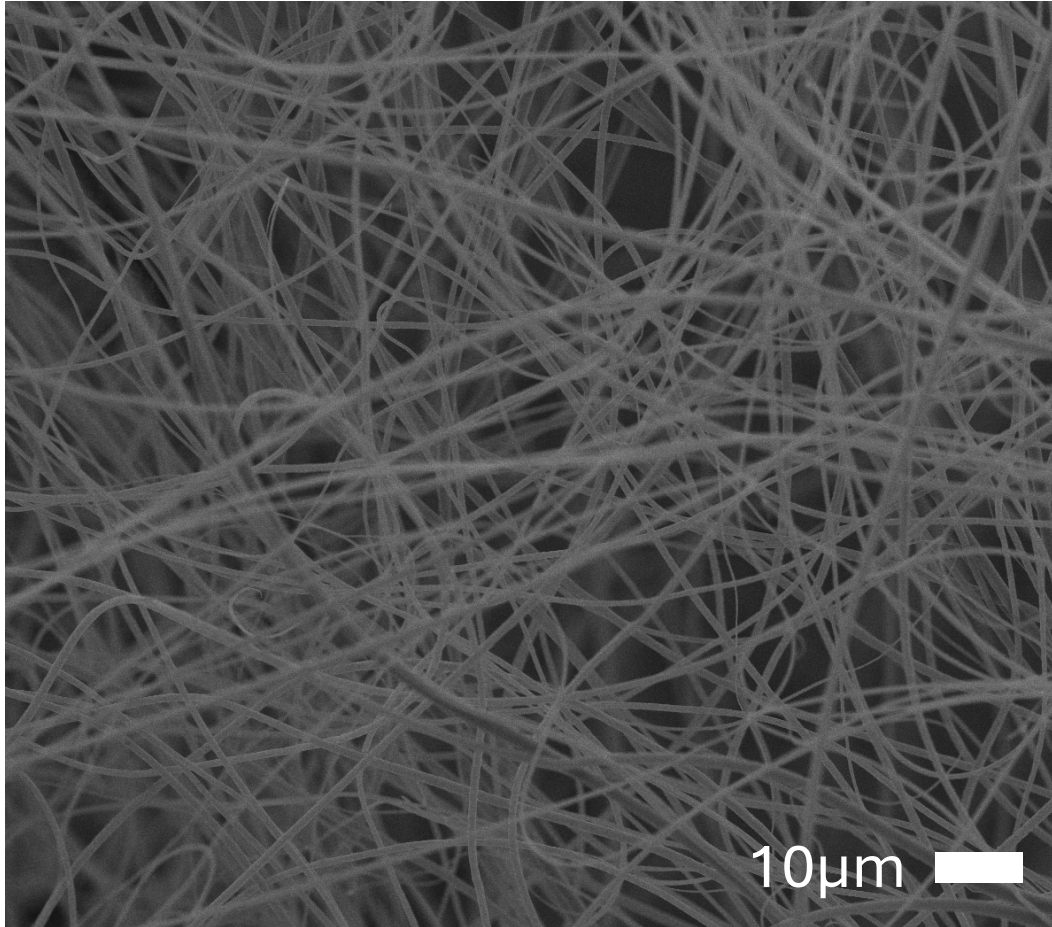
PAN fiber



PAN Aerogel

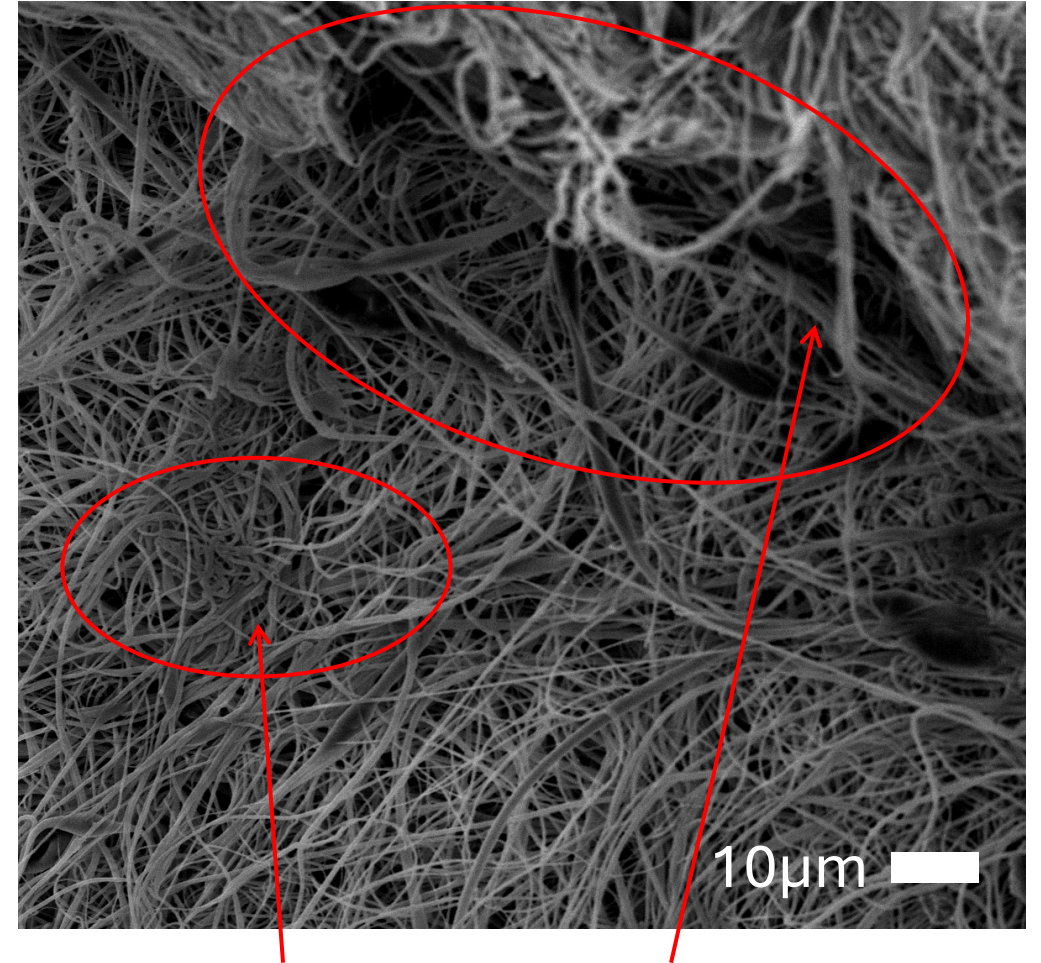
Zoomed Out Aerogel Pictures

Cellulose fiber



Aerogels look denser at nm scale, but are more open/porous as micrometer scale

Cellulose Aerogel



Dense fiber
at pore wall

Large opening/unevenness
at microscale