NNCI—Nanoscale Science and Engineering Exploring Shape Memory Alloys

Explore!

- 1. Take a piece of copper or aluminum wire.
- 2. Twist your wire and then drop it in the hot water. What happens?
- 3. Next take the other wire, twist it, and drop it in the hot water. WATCH!
- 4. What happens?
- 5. Do you want to do it again?



Nitinol is unique because it has two different crystal structures that change phase (in the solid state) at a distinct temperature. When it is below the "transition temperature," it can be stretched and transformed without permanent damage. After the alloy has been reshaped or stretched and heated above its transition temperature, the alloy "recovers" and returns to the unstretched shape.





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Image at: http://en.wikipedia.org/wiki/Nickel_titanium

Applications:

Crystal structures of the atoms of nitinol -

- 1. Martensite: cool form that bends easily
- 2. Austenite: warm, rigid form.

Heat supplies energy to the atoms to move within the structure - solid state phase transformation.



